

# SAFETY DATA SHEET

#### SECTION 1: Identification of the substance/mixture and of the company/undertaking

#### **1.1. Product identifier**

Trade name Sulphur dioxide Product no.

REACH registration number 01-2119485028-34-XXXX Other means of identification EC# 231-195-2, CAS# 7446-09-5

1.2. Relevant identified uses of the substance or mixture and uses advised against

#### Relevant identified uses of the substance or mixture

Sulfur dioxide used in the production of foundry cores, in the paper and pulp industry, in the sugar and starch industry, in the production of pharmaceutical products, in industrial water treatment, in glass coating/lubricate rollers in glass manufacture, in metal casting/mining/purification.

In winemaking, refilling of refrigeration equipment.

# Uses advised against

#### 1.3. Details of the supplier of the safety data sheet

Company and address

Boliden Commercial Box 750 SE-101 35 Stockholm Sweden

Tel +46 8 610 15 00 Fax +46 8 31 55 45 Contact person

#### E-mail

info.market@boliden.com SDS date 19-05-2016 SDS Version 1 1

#### 1.4. Emergency telephone number

999 (or 111 for non-emergency medical advice). Emergency Action: In the event of a medical enquiry involving this product, please contact your doctor or local hospital accident and emergency department or the NHS enquiry service). See section 16.

#### **SECTION 2: Hazards identification**

#### 2.1. Classification of the substance or mixture

Comp. Gas; H280 Acute Tox. 3; H331 Skin Corr. 1B; H314

See full text of H-phrases in section 2.2. 2.2. Label elements

Hazard pictogram(s)





#### **SECTION 3: Composition/information on ingredients**

#### 3.1. Substances

NAME: DENTIFICATION NOS.: 20-9	sulphur dioxide CAS-no: 7446-09-5 EC-no: 231-195-2 REACH-no: 01-2119485028-34-XXXX Index-no: 016-011-
CONTENT:	100%
CLP CLASSIFICATION:	Comp. Gas, Acute Tox. 3, Skin corr. 1B H280, H314, H331

#### 3.2. Mixtures

(\*) See full text of H-phrases in section 16. Occupational exposure limits are listed in section 8, if these are available. **Other informations** 

### **SECTION 4: First aid measures**

#### 4.1. Description of first aid measures

#### **General information**

Get medical advice immediately. Keep warm. Do not scrub the frostbites.

# Inhalation

Remove victim to fresh air and keep at rest in a position comfortable for breathing. Give oxygen or kiss of life.

Skin contact



Remove contaminated clothing and shoes at once. Skin that has come in contact with the material must be washed thoroughly with water.

Get medical advice.

# Eye contact

Remove contact lenses. Flush eyes with water  $(20-30 \,^\circ C)$  for at least 15 minutes. Rinse also under eyelids. Call a doctor.

#### Ingestion

Give the person plenty of water to drink and stay with the person. Contact a doctor immediately and take this safety data sheet or the label from the product with you. Do not induce vomiting. Hold head facing down so that no vomit runs back into the mouth and throat.

#### 4.2. Most important symptoms and effects, both acute and delayed

Sulphur dioxide gas irritates eyes, damp skin areas and airways causing stinging, tearing, cough and at high concentrations breathing difficulties. Spilling of liquid sulphur dioxide may cause frostbites on skin and corneal opacity in eyes. Sulphur dioxide water solutions corrode skin and eyes. Long-term exposure to sulphur dioxide may expose pulmonary diseases, cause bronchitis and damages of tooth enamel.

#### 4.3. Indication of any immediate medical attention and special treatment needed

First aid, decontamination, treatment of symptoms. Follow the advice given in section 4.1.

IF exposed or concerned: Get immediate medical advice/attention.

#### Information to medics

Bring this safety data sheet.

#### **SECTION 5: Firefighting measures**

#### 5.1. Extinguishing media

Use water as a fire extinguishing media. Containers are cooled with water spray. Do not spray water to leakage point.

#### 5.2. Special hazards arising from the substance or mixture

During heating poisonous gases  $(SO_x)$  are evaporated.

#### 5.3. Advice for firefighters

Use compressed air line breathing apparatus and chemical suit.

#### 5.4. Additional information

Intervention actions - General

PUBLIC SAFETY HAZARD - Warn people to stay indoors with doors and windows closed. Stop any ventilation. Consider evacuation of people in immediate danger. Keep upwind. Put in protective equipment before entering danger area. Minimise number of personnel in risk area. Warn people to leave and not to re-enter basements, sewers or other confined spaces.

Intervention actions - Fire (involving the substance)

May react in a fire to produce toxic or irritant gases or fumes. Heating of container(s) will cause pressure rise with risk of bursting and immediate release of expanding toxic and corrosive vapour cloud creating a pressure wave. Contact with liquid will cause frost-bite and severe damage to eyes.

May attack metals and produce hydrogen gas which may form explosive mixtures with air. The gas may be invisible and may enter sewers, basements or confined spaces. Keep container(s) cool with water. Work from protected position to reduce risk to personnel. Use unmanned monitors or lances. Use water-spray to knock down fire fumes if possible.

Avoid unnecessary run-off of extinguishing media which may cause pollution.

#### **SECTION 6: Accidental release measures**

#### 6.1. Personal precautions, protective equipment and emergency procedures

Evacuate leakage area and keep people on the upwind side. Avoid splashes to skin or eyes. Use personal protection. At low concentrations use respiratory protection with E or C type filter. To avoid slashes use heat-insulating rubber gloves, boots and apron. Do not spray water on the liquid.

#### 6.2. Environmental precautions

Close the leakage if possible. Prevent liquids to spray into air with a PVC or PE cover. Gas clouds can be directed (limited) with water-spray.

#### 6.3. Methods and material for containment and cleaning up



Liquid is pumped and collected in slightly opened covered containers or closed pressure-proof containers (evaporation of gas may break the container). Let liquid evaporate in a controlled way.

#### 6.4. Reference to other sections

See section 13 with regard to the handling of waste. See section 8 for protective measures.

#### **SECTION 7: Handling and storage**

#### 7.1. Precautions for safe handling

See section 8 for information on personal protection.

Open and handle the container carefully according to local and national pressure vessel legislation. Use personal protection.

Use only in well ventilated spaces.

Equip the working area with emergency shower and eye rinsing bottle.

# 7.2. Conditions for safe storage, including any incompatibilities

Store in a dry, cool and well-ventilated place. Protect from heat and direct sunshine. Store in the original package.

Incompatible products: Cr, Mn, Al, Zn, Na FeO, SnO, halogenated compounds. Oxygen, oxidising compounds, catalytic materials (to avoid formation of sulphur trioxide).

Hydrogen is formed when sulphur dioxide is reacting with metals in the presence of water. Hydrogen may form an explosive mixture with air.

# Storage temperature

#### No data available.

#### 7.3. Specific end use(s)

This product should only be used for applications described in Section 1.2

#### SECTION 8: Exposure controls/personal protection

#### 8.1. Control parameters

#### OEL

No data available DNEL / PNEC

DNEL

DNEL inhalation (long - term): 0.5 ppm (1.3 mg sulphur dioxide/m3) DNEL inhalation (acute effects): 1 ppm (2.7 mg sulphur dioxide/m3)

The relevant DNELs are based on the recommendations of the "Scientific Committee on Occupational Exposure Limits" (SCOEL) for occupational exposure limits for sulphur dioxide (short-term exposure limit (STEL, 15 min) of 1.0 ppm (2.7 mg/m3) and 8-hour time weighted average (TWA) of 0.5 ppm (1.3 mg/m3)). Guidance on how to comply with these DNELs is given in the attached Exposure Scenarios, in the annex.

PNEC

The gaseous substance SO2 as such does not occur in the aquatic environment. Therefore a PNEC expressed as mg SO2/L is not relevant

#### 8.2. Exposure controls

No control is necessary if the product is used in a normal way.

#### **General recommendations**

Observe general occupational hygiene. Avoid exposure to skin, eyes and clothes.

#### **Exposure scenarios**

If there is an appendix to this safety data sheet, the indicated exposure scenarios must be complied. **Exposure limits** 

There are no maximum exposure limits for the substances contained in this product. **Appropriate technical measures** 

Take ordinary precautions when using the product. Avoid inhalation of gas or dust. **Hygiene measures** 



Whenever you take a break in using this product and when you have finished using it, all exposed areas of the body must be washed. Always wash hands, forearms and face. Measures to avoid environmental exposure

### Individual protection measures, such as personal protective equipment



#### Generally

Only CE-marked personal protection equipment should be used.

#### **Respiratory Equipment**

Use respiration protector if the ventilation is inadequate. At low concentrations use a respiration protector with E or C type filter. At high concentrations use a compressed air line breathing apparatus.

# Skin protection

Use protective clothing

Hand protection Use heat-insulating gloves.

# Eye protection

Use an eye or face protection.

#### **SECTION 9: Physical and chemical properties**

9.1. Information on basic physical and chemical properties					
Form	Colour	Odour	рΗ	Viscosity	Vapour density
Compressed,	-	Pungent	<1	0.25 mPa	about 2,7 kg/m3
liquefied gas				(20°C liquid)	gas
Molting point (%)	N N	Boiling point ( $^{\circ}$ C)			<b>_</b>
-75	)	-10		330 kPa (20 °C)	5
Data on fire and explosion hazard		ls			
Flashpoint (°C)	oint (°C)Ignition (°C)Self ignition (°C)		Self ignition (℃	)	
Explosion limits (	Vol %)	Oxidizing properties			
Solubility		-			
Solubility in wate	r	n-octanol/water coefficient			
Soluble					
113 g/l (20℃)		-			
9.2. Other information	l				
Solubility in fat		Additional information			
		Odour threshold: 1-3 ppm (3-8 mg/m3)			
		Flammability (solid loas): –			
		Relative density: 1380 kg/m3 liquid (20°C)			
_		Decomposition temperature: –			
-					
		Sulfur dioxide is classified as "Liquefied gas - Contains gas under			
		pressure; may explode il nealed. Sulfur dioxide beavier than air: Belative density (air – 1): 2.27: Critical			
		temperature T(c) [ $^{\circ}$ C]: 157.5 $^{\circ}$ C; Critical pressure p(c) [bar]: 78.84 bar			bar]: 78.84 bar
					-

#### **SECTION 10: Stability and reactivity**

- 10.1. Reactivity
- See section 10.5
- 10.2. Chemical stability



Can also position 10.0				
See also section 10.6				
10.3. Possibility of nazardous reactions	5			
See also section 10.5 and 10.6				
10.4. Conditions to avoid				
High temperatures				
10.5. Incompatible materials				
Violent reaction with: ammonia; oxid	ising agents, st	rong chlorine; alkali	is.	
On contact with water: Sulphuric acie	d and sulphuro	us acid.		
Strongly corrosive to metals when m	ioisture is prese	ent.		
10.6. Hazardous decomposition produc	sts			
During heating poisonous gases, su	lphur oxides (S	Ox) are evaporated		
SECTION 11: Toxicological information				
11.1. Information on toxicological effect	te			
	15			
Acute toxicity				
Substance	Species	Test	Route of exposure	Result
Sulphur dioxide	Rat	LC50	Inhalation/1h	2520 ppm
Sulphur dioxide	Human	LCLo	Inhalation/5min,	3000 ppm
Mortal danger (numan)				
30-60 min. 50 - 100 ppm				
5 min. 400 - 500 ppm				
Skin corrosion/irritation				
Gauses severe skin burns.	9			
Liquid sulphur dioxide causes frostbl	ites.			
Serious eye damage/irritation				
Causes serious eye damage.				
Liquid sulphur dioxide causes corner	al damage, cori	neal opacity and blu	ndness.	
Gaseous sulphur dioxide irritates an	d conjunctivas a	and corneal Inflamn	nation.	
Ingestion				
Swallowing corrodes mouth, throat,	stomach and sr	nall intestine.		
Respiratory or skin sensitisation				
No data available.				
Germ cell mutagenicity				
No data available.				
Carcinogenicity				
No data available.				
Reproductive toxicity				
No data available.				
STOT-single exposure				
No data available.				
STOT-repeated exposure				
No data available.				
Aspiration hazard				
Breathing of large concentrations of	sulphur dioxide	e may cause pulmor	nary edema.	
Long term effects				
Tissue damaging effects: This produ	uct contains sul	bstances which are	corrosive. If vapour or ac	erosols are
inhaled, it can result in damage to lu	ngs, irritation a	nd burns in the resp	biratory organs as well as	coughing.
Corrosive substances cause irrevers	sible damage to	eyes and acid burr	ns to skin.	
Other health related data				
Gaseous sulphur dioxide causes cou	ugh, sneeze. te	aring, nausea short	ness of breath. weakness	
unconsciousness and suffocation an	id it irritates nos	se and throat.		
Asthmatic people may get symptoms	s even at low co	oncentrations		
In case of leakage forming sulphur d	lioxide cloud m	av be life-threatenin	I <b>O</b> .	
			.9.	

# **SECTION 12: Ecological information**

**12.1. Toxicity** May lower the pH in waterways and thus be dangerous to live organisms.



Substance No data available.	Species	Test	Test duration	Result
Substance	Biodegradabili Poorly biologically to SO3.	ty degradable. At air oxidizes	Test	Result
12.3. Bioaccumulative potential				
Substance	Potential bioac	ccumulation late.	LogPow	BFC
<b>12.4. Mobility in soil</b> Soluble in water.				
12.5. Results of PBT and vPvB asse The PBT and vPvB criteria do n 12.6. Other adverse effects	<b>ssment</b> ot apply to inorga	anic substances.		
no special				

#### **SECTION 13: Disposal considerations**

### 13.1. Waste treatment methods

The product is covered by the regulations on dangerous waste. Get information of reuse and recycling from the producer. Empty containers are returned to the producer. Dissolve sulphur dioxide into water and neutralize. Dispose of it acccording to national legislation.

Waste

EWC code

Specific labelling

#### **Contaminated packing**

Packaging which contains leftovers from the product must be disposed of in the same way as the product.

# **SECTION 14: Transport information**

This product is covered by the conventions on dangerous goods.

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14.1 -	- 14	4.4	
AD	R/R	ID	

14.1. UN number 14.2. UN proper shipping name 14.3. Transport hazard class(es)	1079 SULPHUR DIOXIDE 2
14.4. Packing group	-
Notes	-
Tunnel restriction code	C/D
IMDG	
UN-no.	1079
Proper Shipping Name	SULPHUR DIOXIDE
Class	2
PG*	-
EmS	-
MP**	-
Hazardous constituent	-
ΙΑΤΑ/ΙCΑΟ	
UN-no.	1079
Proper Shipping Name	SULPHUR DIOXIDE
Class	2
PC*	-
FM	-

14.5. Environmental hazards

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#### 14.6. Special precautions for user

Hazard code 268

14.7. Transport in bulk according to Annex II of MARPOL73/78 and the IBC Code

No data available

(\*) Packing group

(\*\*) Marine pollutant

#### **SECTION 15: Regulatory information**

15.1. Safety, health and environmental regulations/legislation specific for the substance or mixture

#### **Restrictions for application**

People under the age of 18 must not be exposed to this product cf. Council Directive 94/33/EC. Only for industrial use.

**Demands for specific education** 

#### **Additional information**

#### -Sources

EC regulation 1907/2006 (REACH) Directive 2000/532/EC EC Regulation 1272/2008 (CLP)

15.2. Chemical safety assessment

Yes

#### **SECTION 16: Other information**

#### Full text of H-phrases as mentioned in section 3

- H280 Contains gas under pressure; may explode if heated.
- H314 Causes severe skin burns and eye damage.
- H331 Toxic if inhaled.

#### The full text of identified uses as mentioned in section 1

#### Other symbols mentioned in section 2

#### Other

It is recommended to hand over this safety data sheet to the actual user of the product. Information in this safety data sheet cannot be used as a product specification.

The information in this safety data sheet applies only to this specific product (mentioned in section 1) and is not necessarily correct for use with other chemicals/products.

A change (in proportion to the last essential change (first cipher in SDS version)) is marked with a blue triangle.

#### **Emergency Numbers**



Austria: Poison Control Centre Emergency helpline +43 1 406 43 43, 112 Belgium: 070 - 245 245 Bulgaria: +359 2 9154 409 Czech Republic: Toxikologické informační středisko Telefon: +420 224 919 293, +420 224 915 402 Denmark: Kontakt Giftlinien på tlf.nr.: 82 12 12 12 (åbent 24 timer i døgnet). Estonia: 112, 16662, ((+372) 626 93 90) Finland: 09-4711/Myrkytystietokeskus tai suora numero 09-471977 Myrkytystietokeskus/HUS, Tukholmankatu 17, 00029 HUS (Helsinki) 112 France: ORFILA (INRS) : + 33 (0)1 45 42 59 59. 24 heures sur 24 et 7 jours sur 7 Germany: Giftnotruf Berlin, Emergency telephone: +49 30 19240 (Tag und Nacht) Greece: +30 10 779 3777 Hungary: Telefon: 06-80-20-11-99 Iceland: Neyðarlínan: Sími 112. Eitrunarmiðstöð Landsspítalans. Sími: 543 2222. Ireland: +353 1 8379964 Italy: Centro antiveleni di Roma - Policlinico Umberto I tel. 06-49978000 Latvia: +371 704 2468 Lithuania: Visuomenės sveikatos centrams +370 5 236 20 52 arba +370 687 53378 Malta: 2425 0000 Netherlands: 30-2748888 Norway: Giftinformasjonssentralen på tlf.nr.: 22 59 13 00, 113 Poland: +48 58301 65 16 / +48 58 349 2831 Portugal: Em caso de intoxicacao, ligue 808 250 143 Romania: +40 21 3183606 Slovakia: +421 2 54 77 4166 Slovenia: + 386 41 650500 Spain: Servicio de Información Toxicológica Teléfono: + 34 91 562 04 20 (solo emergencias toxicológicas) Información en español (24h/365 días) Sweden: 112, 08-331231 (vardagar kl 9-17) United Kingdom: 999 (or 111 for non-emergency medical advice). Emergency Action: In the event of a medical enquiry involving this product, please contact your doctor or local hospital accident and emergency department or the NHS enquiry service) Date of last essential change (First cipher in SDS version) 01-06-2015 Date of last minor change (Last cipher in SDS version) 19-05-2016

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